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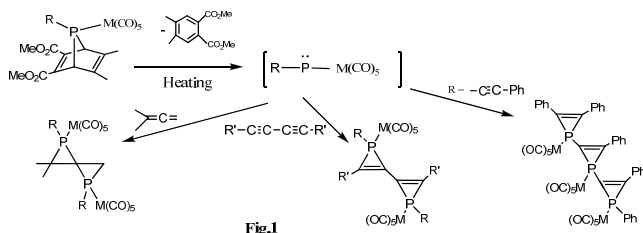
Major Research Interests: Organophosphorus materials for optoelectronic applications; Chemistry of Phosphinidene complexes: carbene like chemistry of monovalent phosphorus.

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1. Electrophilic terminal phosphinidene complexes:

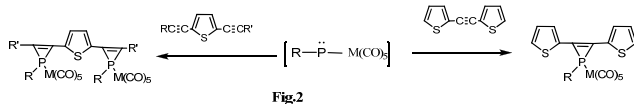
Synthesis of exotic structures

These deficient electron species described here are thermally generated *in situ* from their corresponding 7-phosphanorbornadiene complexes. They behave as singlet carbenes and have a very rich chemistry¹. We report here, only one from its multiple reactivities: their [1+2] cycloaddition with alkenes and alkynes leading to the 3-membered rings phosphirane and phosphirene complexes^{2,3,4}. (Fig.1).



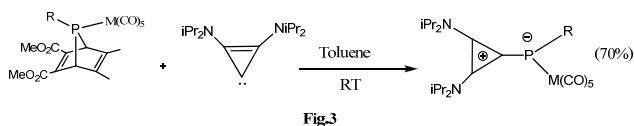
Optoelectronic design

We used phosphirene as conjugating spacer in thiophene oligomers⁵ (Fig.2) to help modulate properties of polythiophenes. We are still far from useful optoelectronic polymer but nevertheless "this synthesis is the first and key step for developing π -conjugated organic materials", as reported in *Chemical & Engineering News* (2006, 84, 12).



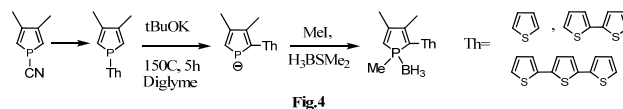
Umpolung of electrophilic terminal phosphinidene complexes by interaction with nucleophilic carbenes

We indeed isolated a stable phosphinidene-carbene 1:1 adduct with a nucleophilic phosphorus⁶ (Fig.3). We are currently exploring the synthetic possibilities of these nucleophilic species.



2. Phosphole-Thiophene Oligomers for optoelectronic applications:

We used the phosphorus 5-membered ring, the phosphole, to modulate the electronic band gap of polythiophenes. By an easy way, we succeeded to get a novel family of Phosphole-Thiophene Oligomers⁷ (Fig. 4). For example, the 1-methyl-2-terthienylphosphole-borane fluoresces at 510nm in CH₂Cl₂ with a quantum yield of 0.17 while the terthiophene fluoresces at 420nm with a quantum yield of 0.06.



References:

1. F. Mathey, N.Hoa Tran Huy, A. Marinetti, *Helv. Chim. Acta* **2001**, 84, 2938.
2. N. Hoa Tran Huy, R. Salemkour, N. Barte, L. Ricard, F. Mathey, *Tetrahedron* **2002**, 58, 7191.
3. N. Hoa Tran Huy, L. Ricard, F. Mathey, *J. Chem. Soc. Dalton Trans.* **1999**, 2409.
4. N. Hoa Tran Huy, L. Ricard, F. Mathey, *Angew. Chem. Int. Ed.* **2001**, 40, 1253.
5. N. Hoa Tran Huy, E. Perrier, L. Ricard, F. Mathey, *Organometallics*, **2006**, 25, 5176.
6. N. Hoa Tran Huy, B. Donnadieu, G. Bertrand, F. Mathey, *Chem. Asian J.* **2009**, 4, 1225.
7. N. Hoa Tran Huy, B. Donnadieu, F. Mathey, A. Muller, K. Colby, C.J. Bardeen, *Organometallics* **2008**, 27, 5521.